

1. Give two examples of solid Sol and Gel? (1)
2. What is the optimum temperature and pH for enzyme catalysed reactions? (1)
3. What are positive and negative catalysts? Explain. (1)
4. Why do finely divided solids act as good adsorbents? (1)
5. Give an equation showing variation of extent of adsorption with concentration of a solution? (1)
6. Give two differences between adsorption and absorption? (2)
7. Write the four differences between physisorption and chemisorption? (2)
8. Define the terms – catalysis and catalyst. (2)
9. Give two examples of enzyme catalysed reaction. (2)
10. Differentiate between multimolecular and macromolecular colloid? (2)
11. Differentiate between lyophobic and lyophilic sol? (3)
12. Explain the terms with suitable examples: (i) Alcosol (ii) Aerosol (iii) Hydrosol (3)
13. What do you mean by activity and selectivity of catalysts? (3)
14. How are colloids classified on the basis of  
(i) Physical states of components (ii) Nature of dispersion medium and  
(iii) Interaction between dispersed phase and dispersion medium? (3)
15. Why is adsorption always exothermic? (3)
16. What is an adsorption isotherm? Describe Freundlich adsorption isotherm. (5)
17. How are the colloidal solutions classified on the basis of physical states of the dispersed phase and dispersion medium? (5)